

## 2.078 Molecular Insights into NO-Promoted Sulfate Formation on Model TiO<sub>2</sub> Mineral Dust with Different Exposed Facets.

Early Career Scientist

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Abstract:

Industry exhaust (such as that from coal-fired plants)-initiated, sulfate aerosol formation on man-made mineral dust has not received a great deal of attention despite having a possible impact on the local environment. Titanium dioxide (TiO<sub>2</sub>) nanoparticle is one of the most widely used materials in the purification of industrial exhaust. Nowadays, there is a lot of efforts, such as through morphology control, have been made to improve the performance of TiO<sub>2</sub> in the application, thus leading to unanticipated release of nano TiO<sub>2</sub> inside the operational areas or outside the nearby atmosphere.

In this study, we choose the most available engineered TiO<sub>2</sub> nanomaterials with preferentially exposed (001), (010) and (101) facet as model oxides, to explore the possible reaction process and mechanism when they encounter NO and SO<sub>2</sub> under typical working conditions. DRIFTS spectra coupled with DFT calculations were used to investigate the dynamic adsorption behaviors of NO or/and SO<sub>2</sub> on the three facets while ion chromatography was adopted to quantify the surface products. Within the operational window of 100-400°C inside the coal-fired plants, an obvious promotion effect of NO on the formation of sulfate species was found on all the faceted samples. Active oxygen species present in those low-coordinated surfaces play a determinant role in the promotion process, which enabled the oxidation of NO into NO<sub>2</sub>. Then the adsorbed nitrate species resulting from the disproportionation of NO<sub>2</sub> dimer (N<sub>2</sub>O<sub>4</sub>) or gaseous NO<sub>2</sub> oxidize adsorbed sulfite into sulfate species. The T101, as the most stable facet existing in nano anatase TiO<sub>2</sub> dust with over 90% fraction, obtained the highest normalized amount of sulfate in the presence of NO. The occurrence of NO-promoted formation of sulfate under simulated atmospheric conditions (30°C, 5 ppmv SO<sub>2</sub> and 10 ppmv NO) indicate that this promotion effect can be ubiquitous on well-engineered anthropogenic dusts.